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different numbers of neutrons are called _____. answer choices . radioactive elements. metalloids. metals.

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properties and increasing atomic numbers is called the periodic table. 3. The mass of an electron is about equal to the mass of a proton. 4. Elements arranged in vertical columns in the periodic table are called periods. 5. The region around the nucleus occupied by the electrons is called the negative zone.

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6. The symbol for chlorine is Cl. 7.

**001 052 C16 CR 896260 3/24/10
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Electrons are considered valence electrons if they are in the outermost electron shell. The energy of electrons decreases with increasing distance from the nucleus. Electrons circle the nucleus

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Chapter 5 TEST: The Periodic Table
Multiple Choice Identify the choice that best completes the statement or answers the question. ____ 1. The order of elements in the periodic table is based on a. the number of protons in the nucleus. b. the electric charge of the nucleus. c. the number of neutrons in the nucleus. d. atomic mass. ____ 2.

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interpretation and essays. Atoms and the Periodic Table includes the following concepts: • Atomic Structure • Masses of Atoms • The Periodic Table • Metals, Nonmetals and Metalloids. Students will be able to:

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A is a group of two or more atoms held together by chemical bonds.

MOLECULE A is a pure substance made of two or more elements chemically combined in a set ratio.

COMPOUND 7. A is a single kind of matter that is pure, meaning it always has a specific makeup, and a specific set of properties and is the same

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All atoms of an element are identical in size, mass, and other properties; atoms of different elements differ in size, mass and other properties. Atoms cannot be subdivided, created, or destroyed.

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Atoms of different elements combine in simple whole-number ratios to form chemical compounds.

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what an element is through the lessons presented in this chapter.

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