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Chemistry Chapter 2 Assessment Answers could include reddish-yellow color, conductor of heat and electricity, malleable, melting point of 1084°C, and boiling point of 2562°C. List three physical properties of copper.

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answer. 157.5 g 106.5 g 51 g Yes. Mass of reactants equals mass of products. Section 3.2 Assessment page 79 10. Classify each example as a physical change or a chemical change. a. crushing an aluminum can physical b. recycling used aluminum cans to make new aluminum cans physical c. aluminum combining with oxygen to form aluminum oxide chemical 11.

Matter—Properties and Changes Matter—Properties and Changes

two acids and two bases to share with the class Acids and Bases Acids & bases are aqueous solutions When acids dissolve in water they form hydrogen ions - H + , and make the solution more acidic Bases form hydroxide ions (OH The chemistry of life chapter 2 assessment answers key. .

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