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GPB 6.34 I. Fill in the blanks with the
most appropriate term:

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Georgia Public Broadcasting**
Chemistry Review Unit 6 - Kinetics /
Equilibrium Kinetics, Equilibrium,

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Spontaneous Reactions Kinetics and Equilibrium 1. Collision theory states that a reaction is most likely to occur if reactant particles collide with the proper energy and orientation.

Chemistry Review - Unit 6 Kinetics

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GPB 6.38a KEY I. Fill in the blanks with
the most appropriate term:

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Sheet 1. On the periodic table below,
identify the following items: Ho '93013
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Unit 14 - Bonding ... Worksheet 2
Worksheet 3 Review Unit 16 -
Equilibrium, Electrochem, Reaction
Rates, Carbon Chemistry Unit 16 ...

Pre-AP Chemistry - Home

are representative of each type of problem worked in this unit. For more practice on a given type of problem, refer back to Note Taking Guides, worksheets, and quizzes.

1. Calculate the molar mass for ethane, C_2H_6 .
 C_2H_6
 $2(12.0 \text{ g}) + 6(1.0 \text{ g}) = 30.0 \text{ g}$

Convert 45 g Se to moles of Se. ? mol Se
 $= 45 \text{ g Se} \div 1 \text{ mol Se } 79.0 \text{ g Se} = 0.57 \text{ mol Se}$

Review Sheet: Unit 7 Name

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Vocabulary: Fill in the blanks ...

Review Sheet: Unit 9 Name _____
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GPB 9.30a KEY FILL IN THE BLANKS and
SHORT ANSWER: 1. The theory that
explains the behavior of ...

FILL IN THE BLANKS and SHORT ANSWER

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that can be your partner.

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Review Sheet: Unit 5 Name _____
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GPB 5.24a KEY I. For each of the
following statements, write "I" for ionic,
"C" for covalent

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Modeling Chemistry 3 54 cm 3 18 g / cm 3 U1 review v2.0 Scientific notation, significant figures, metrics, dimensional analysis and more density (Chapter 5) 1. Represent the following in scientific notation a. 238,000,000 = 2.38×10^8 b. 0.0007548 = 7.548×10^{-4} c. 0.0000045802 = 4.5802×10^{-6} d. 230 = 2.30×10^2 2.

Chemistry - Unit 1 Review

Regents Chemistry. Cumulative Review Materials; Remind101: Assignments and Class Updates; Lab Safety and Equipment; Unit 1: Math and Measurement; Unit 2: Matter and Energy; Unit 3: Behavior of Gases; Unit 4: Atomic Structure; Unit 5: Nuclear Chemistry; Unit 6: Periodic Table; Unit 7: Formulas, Equations and Chemical Reactions; Unit 8: The Mole ...

Piersa, Amanda / Unit 6: Periodic Table

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