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water has a density of 1.00 g/cm<sup>3</sup>. Since a cubic centimeter contains the same volume as a milliliter, in some cases you may see density expressed as g/mL.

Density = volume mass or  $D = \frac{M}{V}$

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Density- Density = Mass divided by Volume ( $D = M/V$ ). The mass of an object is 25 grams. The volume of an object is 5 cm<sup>3</sup>.  $D = 25\text{g}/5\text{cm}^3 = 5 \text{ g/cm}^3$ .

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**Name**

SCIENCE 8 - DENSITY CALCULATIONS

WORKSHEET NAME: 1) A student measures the mass of an 8 cm<sup>3</sup> block of brown sugar to be 12.9 g. What is the density of the brown sugar? 2) A chef fills a 50 mL container with 43.5 g of cooking oil.

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Find the density of a block with a length of 8.0 cm, a width of 4.0cm, a height 2.0 cm, and a mass of 32 g. Would this block float or sink in water?  $V = 8.0 \text{ cm} \times 4.0$

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$4 \text{ cm} \times 2.0 \text{ cm} = 64 \text{ cm}^3$   
 $D = \frac{m}{V} = \frac{32 \text{ g}}{64 \text{ cm}^3} = 0.50 \text{ g/cm}^3$   
c Since the density of the block is less than the density of water, the block would float in water.  
You have two pieces of ...

## **Worksheet: Density Name KEY**

Density is the relationship of the mass of an object to its volume. Density is

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usually reported in units of grams per cubic centimeter ( $\text{g}/\text{cm}^3$ ). For example, water has a density of  $1.00 \text{ g}/\text{cm}^3$ . Since a cubic centimeter contains the same volume as a milliliter, in some cases you may see density expressed as  $\text{g}/\text{mL}$ .

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formula is also a normal density, and  
this will tell you the density of the  
substance. This is more complicated  
than the above one, as you need to

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multiply all of the above numbers with 3.

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